

C(5)—C(6)—C(7)—C(8)] of bufotionin is almost planar within the maximum deviations of 0.04 Å [C(5)].

Fig. 2 depicts the crystal structure projected along the *b* axis. The shortest distance between the positive charge center N(2)⁺ and the negative charge center SO₃⁻ is 3.64 Å [N(2)⋯O(3)(1 - *x*, $\frac{1}{2}$ + *y*, $\frac{1}{2}$ - *z*)]; two methyl groups, C(11) and C(12), block intermolecular interaction between the positive and negative charge centers. A short intermolecular distance of 2.905 (1) Å between the N(1) and O(3) atoms is observed around the center of inversion, which suggests that the two centrosymmetrically related molecules form a dimer linked by hydrogen bonds. Slight enlargement of the double-bond length to 1.452 (2) Å for S(1)—O(3), compared with those for S(1)—O(2) [1.432 (2) Å] and S(1)—O(4) [1.427 (2) Å] is in good agreement with the formation of the hydrogen bond.

The authors are grateful to Mr N. Fukumoto of the laboratory for measurements of the X-ray fluorescence spectrum of the compound.

References

- AKIZAWA, T., OHTANI, K., KASAI, R., GOTO, J. & YOSHIOKA, M. (1990). 39th Annu. Meet. Jpn Soc. Anal. Chem., Nagoya, Japan, 16 October. Abstract 1P17, p. 527.
- B. A. FRENZ & ASSOCIATES INC. (1982). *SDP Structure Determination Package*. College Station, Texas, USA.
- FALKENBERG, G. (1972). *Acta Cryst.* **B28**, 3219–3228.
- JOHNSON, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- MAIN, P., FISKE, S. J., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCQ, J.-P. & WOOLFSON, M. M. (1982). *MULTAN11/82. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Univs. of York, England, and Louvain, Belgium.
- ROBINSON, B., SMITH, G. F., JACKSON, A. H., SHAW, D., FRYDMAN, B. & DEULOFEU, V. (1961). *Proc. Chem. Soc.* pp. 310–311.

Acta Cryst. (1991). **C47**, 1508–1512

Proline Conformations in Linear Peptides. Structure Determination of the Methyl Ester of *N*-Benzyloxycarbonyl-L-prolyl-D-alanine (*N*-Z-L-Pro-D-Ala-OMe)

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(Received 1 October 1990; accepted 2 January 1991)

Abstract. *N*-Z-L-Pro-D-Ala-OMe, C₁₇H₂₂N₂O₅, *M_r* = 334.38, crystallizes in the orthorhombic space group *P*₂₁₂₁, with the cell dimensions *a* = 5.005 (5), *b* = 17.690 (9) and *c* = 18.70 (1) Å³, *V* = 1656.3 Å³, *Z* = 4, *D_x* = 1.341 g cm⁻³, Cu *K*α, λ = 1.54184 Å, μ = 7.830 cm⁻¹, *F*(000) = 712, *T* = 198 K, final *R* (on *F*) = 0.036 for 1575 observed reflections with *I* ≥ 3σ(*I*). The pyrrolidine ring takes on the C₂-C^γ-*endo* conformation. The urethane bond is in the *cis* conformation [ω₀ = 6.0 (3)°] while the peptide bond is in the *trans* conformation [ω₁ = 170.8 (2)°]; φ₁/ψ₁ values are -88.0 (3)° and 151.3 (2)°. Intermolecular hydrogen bonding occurs between the C-terminus and the symmetry-related amide. Systematic examination of the pyrrolidine ring in linear peptides reveals no correlation exists between the *cis*-*trans* orientation of the proline and the conformation of the pyrrolidine ring.

Introduction. Proline is an important constituent of many proteins. Its presence in proteins imposes cer-

tain conformational restrictions, particularly as a helix breaker. Peptides containing proline residues have been extensively studied because of the possibility of *cis*-*trans* isomerization about the X-Pro bond (Carver & Blout, 1976; Grathwohl & Wuthrich, 1976; Nair & Vijayan, 1981) and the different modes of puckering that the pyrrolidine ring can undergo (Balasubramanian, Lakshminarayanan, Sabesan, Tegoni, Venkatesan & Ramachandran, 1971; Ashida & Kakudo, 1974).

Recently it was proposed (Tripathi, Patel & Singh, 1990) that for proline in the *cis* conformation the pyrrolidine ring adopts only the C₂-C^γ-*endo* conformation. The structure of *N*-Z-L-Pro-D-Ala-OMe displays an X-Pro bond which is *cis* and in which the pyrrolidine ring geometry is C₂-C^γ-*endo*. Observation of this ring conformation in conjunction with a *cis* proline appears to further confirm the previously drawn correlations. However, closer examination of a larger database of linear proline-containing peptides reveals the previous assertion to be erroneous.

Experimental. For X-ray examination and data collection, a suitable crystal obtained from ethanol-hexane (approximate dimensions $0.28 \times 0.10 \times 0.13$ mm) was mounted on the tip of a glass fiber. Intensity data were collected at 198 K on an Enraf-Nonius CAD-4 diffractometer with graphite-monochromated Cu $K\alpha$ radiation. Lattice parameters were obtained by least-squares refinement of the angular settings from 25 reflections lying in a 2θ range of $50\text{--}70^\circ$. Intensity data (1746 reflections) were collected using variable speed ω - 2θ scans with $2 \leq 2\theta \leq 135^\circ$ as follows: $0 \leq h \leq 6$; $0 \leq k \leq 21$; $0 \leq l \leq 22$. Three standard reflections ($0\bar{6}9$; $2\bar{9}0$; 275) monitored every 3 h of X-ray exposure time showed negligible nonsystematic intensity changes of $+0.4\%$; no correction for deterioration was made. The data were corrected for absorption effects (correction: min. 0.826, max. 1.244%) based on the *DIFABS* algorithm of Walker & Stuart (1983). The data were also corrected for Lorentz and polarization effects.

The structure was solved by a combination of direct methods with *MULTAN80* (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1980) and the difference Fourier technique and refined by full-matrix least squares (on F). Non-H atoms were refined with isotropic displacement factors, then with anisotropic displacement factors. H-atom positions were located directly from the difference Fourier maps and refined. H-atom isotropic displacement factors were assigned as $1.3 \times B_{\text{eq}}$ of the adjacent atom; they were not refined. The refinement converged [$(\Delta/\sigma)_{\text{max}} = 0.04$] to values of the standard crystallographic agreement factors of $R = 0.036$, $wR = 0.045$ and $S = 1.690$ for 1575 observations with $I \geq 3\sigma(I)$ and 284 parameters. Weights were assigned to the data as $4F^2/\sigma(F^2)$ where $\sigma(F^2) = [\sigma(I)^2 + (0.02F^2)^2]^{1/2}$. An extinction coefficient of the form proposed by Zachariasen (1963) was applied and refined: $g = 1.19 \times 10^{-6}$. Scattering factors were from *International Tables for X-ray Crystallography* (1974, Vol. IV) except for the H atoms (Stewart, Davidson & Simpson, 1965). The effects of anomalous dispersion for non-H atoms were included. A final difference map showed maximum excursions of $(\Delta\rho)_{\text{max}} = 0.177$, $(\Delta\rho)_{\text{min}} = -0.174 \text{ e } \text{\AA}^{-3}$. Final atom positions and equivalent isotropic temperature factors for the non-H atoms are given in Table 1.* All programs used were from the locally modified Enraf-Nonius (1979) *SDP*.

* Lists of anisotropic temperature factors, least-squares planes, torsion angles, H-atom parameters and structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 53881 (22 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. *Positional parameters and e.s.d.'s for N-Z-Pro-Ala-OMe*

$$B_{\text{eq}} = (8\pi^2/3) \sum_j \sum_l U_{jl} a_j^* a_l \cdot a_j \cdot a_l$$

	x	y	z	$B_{\text{eq}}(\text{\AA}^2)$
O8	-0.1493 (4)	0.1419 (1)	0.14108 (9)	2.98 (3)
O9	0.1602 (4)	0.2331 (1)	0.11838 (9)	3.15 (4)
O15	0.2444 (4)	-0.0105 (1)	0.0434 (1)	3.35 (4)
O18	0.0856 (5)	-0.1938 (1)	0.1560 (1)	5.94 (6)
O19	-0.2465 (4)	-0.1129 (1)	0.17691 (9)	3.79 (4)
N10	0.0638 (4)	0.1417 (1)	0.0374 (1)	2.60 (4)
N16	-0.1834 (4)	-0.0511 (1)	0.0446 (1)	2.43 (4)
C1	-0.4195 (5)	0.1319 (1)	0.2461 (1)	2.84 (5)
C2	-0.4855 (6)	0.1513 (2)	0.3159 (2)	3.74 (6)
C3	-0.6840 (7)	0.1123 (2)	0.3521 (2)	4.17 (6)
C4	-0.8189 (6)	0.0537 (2)	0.3191 (2)	4.10 (6)
C5	-0.7533 (7)	0.0339 (2)	0.2497 (2)	4.25 (7)
C6	-0.5526 (6)	0.0726 (2)	0.2135 (2)	3.61 (6)
C7	-0.2093 (6)	0.1777 (2)	0.2087 (1)	3.16 (5)
C9	0.0358 (5)	0.1770 (1)	0.1006 (1)	2.53 (5)
C11	0.2360 (5)	0.1732 (1)	-0.0187 (1)	2.81 (5)
C12	0.2033 (5)	0.1173 (2)	-0.0801 (1)	2.92 (5)
C13	-0.0798 (5)	0.0864 (1)	-0.0696 (1)	2.71 (5)
C14	-0.1020 (5)	0.0789 (1)	0.0126 (1)	2.30 (4)
C15	0.0050 (5)	0.0025 (1)	0.0363 (1)	2.34 (4)
C17	-0.1141 (5)	-0.1297 (1)	0.0567 (1)	2.55 (5)
C18	-0.0752 (6)	-0.1483 (1)	0.1351 (2)	3.08 (5)
C20	-0.2481 (8)	-0.1350 (2)	0.2520 (2)	5.35 (8)
C21	-0.3303 (6)	-0.1810 (2)	0.0270 (2)	3.52 (6)

Discussion. The molecular structure of *N-Z-L-Pro-D-Ala-OMe* is shown in Fig. 1. The stereo unit-cell diagram is shown in Fig. 2. Principal bond distances and angles are given in Table 2. Selected torsion angles are given in Table 3.

The benzyloxycarbonyl-blocked N-terminus (Benedetti, Pedone, Toniolo, Dudek, Nemethy & Scheraga, 1983) and methyl ester-blocked C-terminus (Schweizer & Dunitz, 1982) display bond distances and bond angles consistent with other peptides containing these groups. As is consistent with other proline-containing peptides, the phenyl and proline rings tend to stack in the crystals as is illustrated in Fig. 2.

The urethane bond in *N-Z-Pro-Ala-OMe* is in the *cis* conformation [$\omega_0 = 6.0(3)^\circ$] while the peptide bond is in the *trans* conformation [$\omega_1 = 170.8(2)^\circ$]. The observed values for ω are comparable to the normal values of 0 and 180° for peptide bonds with *cis* and *trans* conformations, respectively (Benedetti, 1982). The urethane group is nearly planar, as seen by comparing the two torsion angles $\theta^1 = \text{C7—O8—C9—N10} = -176.2(2)$ and $\theta^{1'} = \text{C7—O8—C9—O9} = 3.5(4)^\circ$. As is generally found in esters, the C9=O9 bond is synperiplanar to the C7—O8 bond, the $\theta^{1'}$ torsion angle being nearly 0° (Benedetti, Pedone, Toniolo, Dudek, Nemethy & Scheraga, 1983). The principal torsion angles φ_1 and ψ_1 are $-88.0(3)$ and $151.3(2)^\circ$, respectively, and are consistent with other *cis* proline residues.

There is no evidence of disorder in the C^γ atom for *N-Z-Pro-Ala-OMe* although such disorder is frequently observed in proline-containing peptides

(Ashida & Kakudo, 1974). The bond angles of the pyrrolidine ring, particularly about the N atom, are strongly dependent on the *cis-trans* conformation. The *cis* geometry is characterized by a larger $C'-N-C^\alpha$ angle and a smaller $C'-N-C^\delta$ angle as compared to the *trans* conformation (Benedetti, 1982). The differences in angles can be readily observed if *cis-N-Z-Pro-Ala-OMe* [125.0 (2) and 120.9 (2)°, respectively] is compared to *trans-tert-Boc-Pro-Leu-Gly.H₂O* (120 and 124°, respectively) (Ashida, Tanaka, Shimonishi & Kakudo, 1977) or *trans-N-Z-Ala-Pro* [120.1 (6) and 127.3 (5)°, respectively] (Krause & Eggleston, 1990; Panneerselvam, Chacko & Veena, 1990).

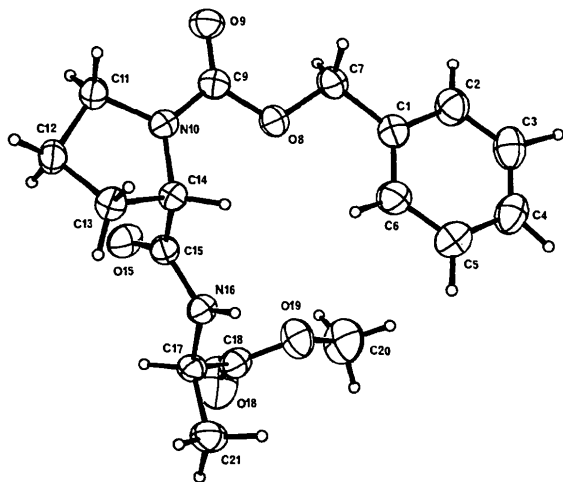


Fig. 1. ORTEP (Johnson, 1976) drawing of *N-Z-Pro-Ala-OMe* showing 50% thermal-ellipsoid probability for the non-H atoms, H atoms as small spheres of arbitrary size and the atomic labelling scheme.

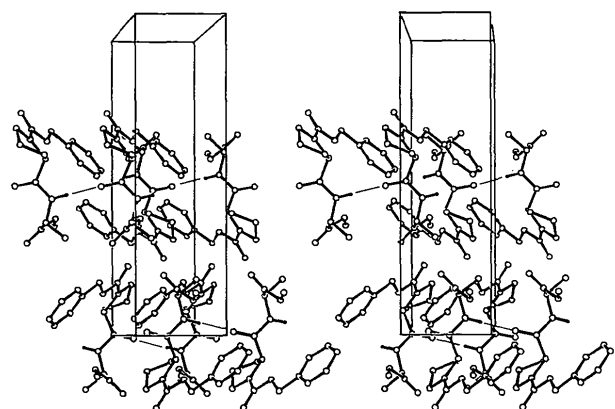


Fig. 2. Stereo unit-cell drawing of *N-Z-Pro-Ala-OMe* indicating the intermolecular hydrogen bonding. Hydrogen-bonding interactions are indicated by single lines. The *a* axis is along the horizontal while the *b* axis runs vertically.

Table 2. Selected bond distances (Å) and bond angles (°) for *N-Z-Pro-Ala-OMe* with *e.s.d.*'s in parentheses

O8—C7	1.446 (2)	C1—C6	1.385 (3)
O8—C9	1.348 (3)	C1—C7	1.501 (3)
O9—C9	1.218 (2)	C2—C3	1.385 (4)
O15—C15	1.228 (3)	C3—C4	1.382 (4)
O18—C18	1.203 (3)	C4—C5	1.383 (4)
O19—C18	1.318 (3)	C5—C6	1.392 (4)
O19—C20	1.458 (3)	C11—C12	1.524 (3)
N10—C9	1.345 (3)	C12—C13	1.531 (3)
N10—C11	1.469 (3)	C13—C14	1.547 (3)
N10—C14	1.463 (3)	C14—C15	1.520 (3)
N16—C15	1.347 (3)	C17—C18	1.516 (3)
N16—C17	1.450 (3)	C17—C21	1.518 (3)
C1—C2	1.390 (3)		
C7—O8—C9	115.6 (2)	O9—C9—N10	124.5 (2)
C18—O19—C20	116.6 (2)	N10—C11—C12	103.2 (2)
C9—N10—C11	120.9 (2)	C11—C12—C13	103.5 (2)
C9—N10—C14	125.0 (2)	C12—C13—C14	103.0 (2)
C11—N10—C14	113.2 (2)	N10—C14—C13	102.1 (2)
C15—N16—C17	121.7 (2)	N10—C14—C15	112.5 (2)
C2—C1—C6	119.1 (2)	C13—C14—C15	110.0 (2)
C2—C1—C7	118.1 (2)	O15—C15—N16	122.6 (2)
C6—C1—C7	122.7 (2)	O15—C15—C14	122.9 (2)
C1—C2—C3	120.4 (3)	N16—C15—C14	114.4 (2)
C2—C3—C4	120.4 (2)	N16—C17—C18	113.0 (2)
C3—C4—C5	119.5 (3)	N16—C17—C21	110.2 (2)
C4—C5—C6	120.2 (3)	C18—C17—C21	108.4 (2)
C1—C6—C5	120.3 (3)	O18—C18—O19	124.1 (2)
O8—C7—C1	108.5 (2)	O18—C18—C17	123.1 (2)
O8—C9—O9	125.0 (2)	O19—C18—C17	112.7 (2)
O8—C9—N10	110.6 (2)		

Table 3. Selected torsion angles (°) and *e.s.d.*'s for *N-Z-Pro-Ala-OMe*

O8—C9—N10—C14	6.0 (3)	ω_0	C14—C13—C12—C11	-39.0 (2)	χ^2
C14—C15—N16—C17	170.8 (2)	ω_1	C13—C12—C11—N10	29.5 (2)	χ^3
C9—N10—C14—C15	-88.0 (3)	φ_1	C12—C11—N10—C14	-9.0 (3)	χ^4
C15—N16—C17—C18	87.7 (3)	φ_2	C11—N10—C14—C13	-15.0 (3)	$\chi = \theta$
C15—N16—C17—C21	-150.8 (2)	χ_2	C15—C14—C13—C12	-86.9 (2)	θ^i
N10—C14—C15—N16	151.3 (2)	ψ_1	C15—C14—N10—C11	102.9 (2)	θ^{ii}
N16—C17—C18—O18	-145.8 (3)	ψ_2	C9—N10—C11—C12	-178.6 (2)	θ^{iii}
N16—C17—C18—O19	38.0 (3)	ψ_3	C9—N10—C14—C13	154.1 (2)	θ^{iv}
N10—C14—C13—C12	32.7 (2)	χ^1			

The pyrrolidine ring in *N-Z-Pro-Ala-OMe* is puckered and adopts the C_2-C^γ -*endo* conformation which may be defined as a half-chair with C^β and C^γ residing on opposite sides of the plane defined by $N-C^\alpha-C^\delta$ and C^γ on the same side of this plane as C' of Pro (Ashida & Kakudo, 1974). It has been proposed, based apparently on the analysis of 13 structures, that a correlation exists between the *cis* orientation of the proline and the conformation of the pyrrolidine ring (Tripathi, Patel & Singh, 1990). The structure revealed for *Z-L-Pro-D-Ala-OMe* thus appears to further substantiate this conclusion. The torsion angle values for a number of proline-containing peptides are compiled in Table 4. Systematic examination of the various ω torsion angles, urethane and peptide angles, associated with the proline reveals that no correlation exists between either a *cis* or *trans* orientation of the proline and the conformation of the pyrrolidine ring. For example, in the structure of *Boc-Pro-Val-Gly-NH₂* (Tanaka &

Table 4. Torsion angles ($^{\circ}$) and pyrrolidine ring conformations of selected linear proline residues

For those structures with two independent molecules, values in parentheses are for molecule B.											
Compound	ω_0^*	ω_{x-1}^*	ω_x^*	Conformation†	Ref.	Compound	ω_0^*	ω_{x-1}^*	ω_x^*	Conformation†	Ref.
Z-Pro-Ala-OMe	6.0 (3)		170.8 (2)	C_2-C^{γ} -endo	1	Boc-Pro-Leu-Gly-NH ₂ (δ -lactam)	-10		178	C_2-C^{γ} -exo	17
Boc-Pro-Met-Gly-OBzl	-179		-20	C_2-C^{γ} -endo	2	Boc-Pro-Val-Gly.1/2H ₂ O	8		163	C_2-C^{γ} -endo	18
	(176)		(-13)	(C_2-C^{γ} -endo)			(2)		(154)	(C_2-C^{γ} -endo)	
Boc-Pro-Sar-OBzl	-13		-7	C_2-C^{γ} -endo	3	Boc-Pro-Val	-8			C_2-C^{γ} -exo	19
Z-Gly-Pro-Leu		-4	-177	C_2-C^{γ} -endo	4	Boc-Pro-Val-Gly-NH ₂	0.1		166	C_2-C^{γ} -endo	20
S-Bz-Cys-Pro-Leu-Gly-NHMe		4	4	C_2-C^{γ} -endo	5		(4)		(153)	(C_2-C^{β} -exo)	
		(3)	(1)	(C_2-C^{γ} -endo)		Boc-Pro-HGly-OEt	-11		-178	C_1-C^{β} -exo	21
Se-Bz-Cys-Pro-Leu-Gly-NHMe		0	-1	C_1-C^{β} -exo	5	Boc-Pro-Ala	8		-171	C_2-C^{γ} -endo	22
		(6)	(8)	(C_1-C^{β} -endo)			(4)		(174)	(C_1-C^{γ} -endo)	
Boc-Pro	-6			C_2-C^{γ} -endo	6	Boc-Pro-Ala-Ala	-177		174	C_2-C^{γ} -endo	22
Z-Pro-Ala-Thr('Bu) ₂	3		178	C_2-C^{γ} -endo	7	Boc-Pro-dehydro-Leu-OMe	-2		169	C_1-C^{β} -exo	23
Z-Pro-Leu-OEt	-11		171	C_2-C^{γ} -endo	8		(-3)		(167)	(C_2-C^{γ} -endo)	
Boc-Pro-Leu-OBzl	-13		157	C_2-C^{γ} -endo	8	Boc-Pro-dehydro-Leu-NHMe	176		-179	C_2-C^{γ} -exo	24
Boc-Pro-Pro	-7		-176	C_1-C^{γ} -endo (Pro 1)	9	Boc-Pro-dehydro-Phe-Gly	179		175	C_2-C^{γ} -exo	25
				C_1-C^{γ} -endo (Pro 2)		Boc-Pro-His-NHMe	-178		177	C_2-C^{γ} -exo	26
Boc-Pro-Val-OMe:	3		168	C_1-C^{γ} -endo		Tos-Pro-hydroxy-Pro.H ₂ O	-78		176	C_1-C^{γ} -endo (Pro 1)	27,28
Boc-Pro-C ^β -Me-Val-OMe (1:1)	-1		167	C_1-C^{γ} -endo	10			176		C_2-C^{γ} -exo (Pro 2)	
Boc-Pro-Ala-Gly-NH ₂	-13		-171	C_2-C^{γ} -endo	11	Poc-Pro-Ala-Gly	-3		-179	C_1-C^{β} -exo	29
Boc-Pro-Pro-Gly-NH ₂	-2		-179	C_2-C^{γ} -exo (Pro 1)	12		(3)		(-178)	(C_1-C^{β} -exo)	
		-179	178	C_2-C^{γ} -exo (Pro 2)		Aoc-Pro-Pro-Pro	-10		172	C_1-C^{β} -exo (Pro 1)	30
Boc-Pro ₄ -OBzl.H ₂ O	-1		169	Intermediate (Pro 1)‡	13				-179	C_1-C^{β} -exo (Pro 2)	
			176	C_2-C^{γ} -exo (Pro 2)		Aoc-Pro	-2			C_2-C^{γ} -endo (Pro 3)	
			180	C_1-C^{β} -exo (Pro 3)		Z-Ala-Pro		171		C_1-C^{γ} -endo	31
			180	C_1-C^{β} -exo (Pro 4)		p-Br-Z-Gly-Pro-Leu-Gly	-174	175		C_2-C^{γ} -exo	32,33
Boc-Pro-Sar	-7		170	C_2-C^{γ} -endo	14,15	o-Br-Z-Gly-Pro-Leu-Gly-Pro	-168	-178		C_1-C^{β} -endo (Pro 1)	35
Boc-Pro-Ile-Gly	11		172	C_2-C^{γ} -endo	16		178			C_2-C^{γ} -endo (Pro 2)	
Boc-Pro-Leu-Gly-NH ₂ .1/2H ₂ O	-10		180	C_1-C^{β} -exo	17	Z-Gly-Pro-Leu-Gly-Pro.2H ₂ O	-177	-179		C_1-C^{γ} -exo (Pro 1)	36
	(-13)		(-171)	(C_2-C^{γ} -endo)			181			C_1-C^{γ} -endo (Pro 2)	
Boc-Pro-Leu-Gly-NH ₂ (γ -lactam)	-4		174	C_2-C^{γ} -endo	17	Tyr-Pro-Asp-Gly		176	182	C_2-C^{γ} -endo	37

References: (1) This work; (2) Yamane, Shiraishi & Ashida (1985); (3) Kojima, Kido, Itoh, Yamane & Ashida (1980); (4) Yamane, Ashida, Shimonishi, Kakudo & Sasada (1976); (5) Rudko & Low (1975); (6) Benedetti, Ciajolo & Maisto (1974); (7) Baoguang, Yicheng, Zhengjiong & Youqi (1984); (8) Sugino, Tanaka & Ashida (1978); (9) Kamwaya, Oster & Bradaczek (1981); (10) Trikha, Patel & Singh (1990); (11) Kojima, Tanaka & Ashida (1982); (12) Tanaka, Ashida, Shimonishi & Kakudo (1979); (13) Matsuzaki (1974); (14) Itoh, Yamane & Ashida (1978); (15) Benedetti, Ciajolo, Di Blasio, Pavone, Pedone, Toniolo & Bonora (1979a); (16) Yamada, Tanaka & Ashida (1980); (17) Valle, Crisma, Toniolo, Yu & Johnson (1989); (18) Tanaka & Ashida (1980); (19) Bosch, Schmitt, Jung & Winter (1984); (20) Ashida, Kojima, Tanaka & Yamane (1986); (21) Viret, Collet, Pichon-Pesme & Aubry (1988); (22) Ananthanarayanan & Cameron (1988); (23) Narula, Patel & Singh (1988); (24) Singh, Narula, Chauhan, Sharma & Hinrichs (1989); (25) Patel, Singh, Chauhan & Kaur (1990); (26) Aubry, Vlasi & Marraud (1986); (27) Sabesan & Ventatesan (1971); (28) Fridrichsons & Mathieson (1962); (29) Yamada, Tanaka & Ashida (1981); (30) Kartha, Ashida & Kakudo (1974); (31) Benedetti, Ciajolo, Di Blasio, Pavone, Pedone, Toniolo & Bonora (1979b); (32) Panneerselvam, Chacko & Veena (1990); (33) Krause & Eggleston (1990); (34) Ueki, Ashida, Kakudo, Sasada & Katsube (1969); (35) Ueki, Bando, Ashida & Kakudo (1971); (36) Bando, Tanaka, Ashida & Kakudo (1978); (37) Precigoux, Geoffre, Hospital & Leroy (1982).

* ω_0 refers to urethane bond, ω_{x-1} refers to peptide bond preceding proline, ω_x refers to peptide bond following proline.

† Conformation is based on the proline torsion angles as defined by Ashida & Kakudo (1974).

‡ Virtually planar ring, conformation may be described as C_1-C^{γ} -exo.

Ashida, 1980), in which the urethane group is *cis*, the pyrrolidine ring conformation is C_5-C^{γ} -endo. Likewise, the structure of S-Bz-Cys-Pro-Leu-Gly-NHMe (Rudko & Low, 1975), which contains two independent molecules, both with *cis* X-Pro bonds, has the pyrrolidine rings in both the C_2-C^{γ} -endo and C_5-C^{γ} -endo conformations. Secondly, the conformation of one pyrrolidine ring does not influence the conformation of an adjacent ring, as revealed in the structures of Boc-Pro₄-OBzl (Matsuzaki, 1974) and Aoc-Pro₃-OH (Kantha, Ashida & Kakudo, 1974). Considering the rather modest energy requirements for pyrrolidine ring deformation along the pseudo-rotational pathway and the effects which crystal packing forces could be expected to induce, the lack of such a correlation upon examination of a suitably sized database is not surprising.

The crystal structure of Z-Pro-Ala-OMe displays intermolecular hydrogen bonding between the symmetry-related amide groups [N16...O15 = 2.953 (2), N16—H15 = 0.87 (1), H15...O15 = 2.08 (1) Å and N16—H15...O15 = 179 (1) $^{\circ}$] along the *x* axis as indicated in Fig. 2. No intramolecular hydrogen bonding is observed.

This work was supported in part by grant No. GM39S26-02 from the National Institutes of Health. JAK thanks the NIH for postdoctoral support under this grant. The authors thank Dr K. Kopple (Smith Kline Beecham Pharmaceuticals) for the peptide sample.

References

- ANANTHANARAYANAN, V. S. & CAMERON, T. S. (1988). *Int. J. Pept. Protein Res.* **31**, 399–411.
- ASHIDA, T. & KAKUDO, M. (1974). *Bull. Chem. Soc. Jpn.*, **47**, 1129–1133.
- ASHIDA, T., KOJIMA, I., TANAKA, I. & YAMANE, T. (1986). *Int. J. Pept. Protein Res.* **27**, 61–69.
- ASHIDA, T., TANAKA, I., SHIMONISHI, Y. & KAKUDO, M. (1977). *Acta Cryst.* **B33**, 3054–3059.
- AUBRY, A., VLASSI, M. & MARRAUD, M. (1986). *Int. J. Pept. Protein Res.* **28**, 637–648.
- BALASUBRAMANIAN, R., LAKSHMINARAYANAN, A. V., SABESAN, M. N., TEGONI, G., VENKATESAN, K. & RAMACHANDRAN, G. N. (1971). *Int. J. Pept. Protein Res.* **3**, 25–33.
- BANDO, S., TANAKA, N., ASHIDA, T. & KAKUDO, M. (1978). *Acta Cryst.* **B34**, 3447–3449.
- BAOGUANG, Z., YICHENG, D., ZHENGJIONG, L. & YOUQI, T. (1984). *Sci. Sin.* **B27**, 558–565.
- BENEDETTI, E. (1982). *Chemistry and Biochemistry of Amino Acids, Peptides and Proteins*, Vol. 6, edited by B. WEINSTEIN, pp. 105–118. New York: Marcel Dekker.

- BENEDETTI, E., CIAJOLO, A., DI BLASIO, B., PAVONE, V., PEDONE, C., TONIOLO, C. & BONORA, G. M. (1979a). *Macromolecules*, **12**, 438–445.
- BENEDETTI, E., CIAJOLO, A., DI BLASIO, B., PAVONE, V., PEDONE, C., TONIOLO, C. & BONORA, G. M. (1979b). *Int. J. Pept. Protein Res.* **14**, 130–142.
- BENEDETTI, E., CIAJOLO, M. R. & MAISTO, A. (1974). *Acta Cryst.* **B30**, 1783–1788.
- BENEDETTI, E., PEDONE, C., TONIOLO, C., DUDEK, M., NEMETHY, G. & SCHERAGA, J. A. (1983). *Int. J. Pept. Protein Res.* **21**, 163–181.
- BOSCH, R., SCHMITT, H., JUNG, G. & WINTER, W. (1984). *Acta Cryst.* **C40**, 1096–1098.
- CARVER, J. P. & BLOUT, E. R. (1976). *Treatise on Collagen*, Vol. I, edited by G. N. RAMACHANDRAN. New York: Academic Press.
- Enraf–Nonius (1979). *Structure Determination Package*. Enraf–Nonius, Delft, The Netherlands.
- FRIDRICHSONS, J. & MATHIESON, A. MCL. (1962). *Acta Cryst.* **15**, 569–577.
- GRATHWOHL, C. & WUTHRICH, K. (1976). *Biopolymers*, **15**, 2025–2041.
- ITO, H., YAMANE, T. & ASHIDA, T. (1978). *Acta Cryst.* **B34**, 2640–2643.
- JOHNSON, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- KAMWAYA, M. E., OSTER, O. & BRADACZEK, H. (1981). *Acta Cryst.* **B37**, 1564–1568.
- KARTHA, G., ASHIDA, T. & KAKUDO, M. (1974). *Acta Cryst.* **B30**, 1861–1866.
- KOJIMA, T., KIDO, T., ITOH, H., YAMANE, T. & ASHIDA, T. (1980). *Acta Cryst.* **B36**, 326–331.
- KOJIMA, T., TANAKA, I. & ASHIDA, T. (1982). *Acta Cryst.* **B38**, 221–225.
- KRAUSE, J. A. & EGGLESTON, D. S. (1990). Unpublished results.
- MAIN, P., FISKE, S. J., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCQ, J.-P. & WOOLFSON, M. M. (1980). *MULTAN80. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Univs. of York, England, and Louvain, Belgium.
- MATSUZAKI, T. (1974). *Acta Cryst.* **B30**, 1029–1036.
- NAIR, C. M. K. & VUJAYAN, M. (1981). *J. Indian Inst. Sci.* **63**, 81–103.
- NARULA, P., PATEL, H. C. & SINGH, T. P. (1988). *Biopolymers*, **27**, 1595–1606.
- PANNEERSELVAM, K., CHACKO, K. K. & VEENA, K. R. (1990). *Acta Cryst.* **C46**, 81–84.
- PATEL, H. C., SINGH, T. P., CHAUHAN, V. S. & KAUR, P. (1990). *Biopolymers*, **29**, 509–515.
- PRECIGOUX, G., GEOFFRE, S., HOSPITAL, M. & LEROY, F. (1982). *Acta Cryst.* **B38**, 2172–2176.
- RUDKO, A. D. & LOW, B. W. (1975). *Acta Cryst.* **B31**, 713–725.
- SABESAN, M. N. & VENTATESAN, K. (1971). *Acta Cryst.* **B27**, 1879–1883.
- SCHWEIZER, W. B. & DUNITZ, J. D. (1982). *Helv. Chim. Acta*, **65**, 1547–1554.
- SINGH, T. P., NARULA, P., CHAUHAN, V. S., SHARMA, A. K. & HINRICHS, W. (1989). *Int. J. Pept. Protein Res.* **33**, 167–172.
- STEWART, R. F., DAVIDSON, E. R. & SIMPSON, W. T. (1965). *J. Chem. Phys.* **42**, 3175–3187.
- SUGINO, H., TANAKA, I. & ASHIDA, T. (1978). *Bull. Chem. Soc. Jpn*, **51**, 2855–2861.
- TANAKA, I. & ASHIDA, T. (1980). *Acta Cryst.* **B36**, 2164–2167.
- TANAKA, I., ASHIDA, T., SHIMONISHI, Y. & KAKUDO, M. (1979). *Acta Cryst.* **B35**, 110–114.
- TRIKHA, J., PATEL, J. C. & SINGH, T. P. (1990). *Acta Cryst.* **C46**, 74–78.
- UEKI, T., ASHIDA, T., KAKUDO, M., SASADA, Y. & KATSUBE, Y. (1969). *Acta Cryst.* **B25**, 1840–1849.
- UEKI, T., BANDO, S., ASHIDA, T. & KAKUDO, M. (1971). *Acta Cryst.* **B27**, 2219–2231.
- VALLE, G., CRISMA, M., TONIOLO, C., YU, K.-L. & JOHNSON, R. L. (1989). *J. Chem. Soc. Perkin Trans.* **2**, 83–87.
- VIRET, J., COLLET, A., PICHON-PESME, V. & AUBRY, A. (1988). *New J. Chem.* **12**, 253–256.
- WALKER, N. & STUART, D. (1983). *Acta Cryst.* **A39**, 158–166.
- YAMADA, Y., TANAKA, I. & ASHIDA, T. (1980). *Acta Cryst.* **B36**, 331–335.
- YAMADA, Y., TANAKA, I. & ASHIDA, T. (1981). *Bull. Chem. Soc. Jpn*, **54**, 69–72.
- YAMANE, T., ASHIDA, T., SHIMONISHI, K., KAKUDO, M. & SASADA, Y. (1976). *Acta Cryst.* **B32**, 2071–2076.
- YAMANE, T., SHIRAISHI, Y. & ASHIDA, T. (1985). *Acta Cryst.* **C41**, 946–950.
- ZACHARIASEN, W. H. (1963). *Acta Cryst.* **16**, 1139–1144.

Acta Cryst. (1991). **C47**, 1512–1515

6-Amino-1,3-dimethyl-5-(2-ethylphenylazonio)uracil Bromide Dihydrate

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(Received 31 August 1990; accepted 20 November 1990)

Abstract. $C_{14}H_{18}N_5O_2^+ \cdot Br^- \cdot 2H_2O$, $M_r = 404.3$, triclinic, $P\bar{1}$, $a = 7.1738$ (8), $b = 10.110$ (2), $c = 13.394$ (2) Å, $\alpha = 70.34$ (1), $\beta = 75.61$ (1), $\gamma = 79.34$ (1)°, $V = 880.6$ (2) Å³, $Z = 2$, $D_x = 1.52$ g cm⁻³, graphite-monochromatized Mo $K\alpha$

radiation, $\lambda = 0.71069$ Å, $\mu = 27.3$ cm⁻¹, $F(000) = 416$, $T = 298$ K, $R = 0.052$ for 2640 observed reflections. The bulky organic cation is essentially coplanar and the protonation takes place at the azo nitrogen N(8). The additional proton participates in